Localization of carriers of Y-doped CaCuO₂ films prepared by an organometallic-chemicalvapour-deposition method

KENKICHIRO KOBAYASHI, YOZO ISHIHARA, SHIGENORI MATSUSHIMA, GENJI OKADA Department of Applied Chemistry, Faculty of Engineering, Ehime University, Matsuyama 790, Japan

Y-doped CaCuO₂ films have been formed on SrTiO₃ (100) substrates in the range 730–790 °C. The lattice constants of $c = 0.3180 \sim 0.3183$ nm were estimated from the X-ray diffraction (XRD) of *c*-axis-oriented Y-doped CaCuO₂ films. The values of the *c*-axis were shorter than those of the undoped CaCuO₂ film. Although the resistivity of a Y-doped CaCuO₂ film is significantly lower in comparison with that of an undoped CaCuO₂ film, no superconductivity was found in these Y-doped CaCuO₂ films at temperatures higher than 12 K.

1. Introduction

 $CaCuO_2$ with infinite CuO_2 sheets is the parent material in superconductors of a Bi-Sr-Ca-Cu-O and a Tl-Ba-Ca-Cu-O system. In the CaCuO₂ structure, an appreciable interaction is probably present between the CuO_2 sheets, so that the correlation of the interaction between the CuO_2 sheets to a T_c value may be revealed from investigation of the superconductivity of CaCuO₂. Electron doping of CaCuO₂ has been attempted to obtain a superconducting CaCuO₂. No one has succeeded , however, in the electron doping of CaCuO₂. Recently, Smith et al. [1] found superconductivity at 40 K in Nd-doped $SrCuO_2$, with infinite CuO_2 sheets, which was prepared by means of a solid-state reaction at high pressure. They interpreted the failure of electron-doping of CaCuO₂ as being due to a small Cu-O bond length as in Gd_2CuO_4 [1]. Another reason may be associated with the instability of CaCuO₂ at high pressure, because CaCuO₂ has not been prepared at high pressures [2]. Using an organometallic-chemicalvapour-deposition (MOCVD) method, the authors succeeded in the preparation of $CaCuO_2$ [3]; this was not synthesized by the conventional solid-state reaction. The present work attempted the preparation of Y-doped $CaCuO_2$ films by employing the MOCVD method, and the electrical properties of the Y-doped CaCuO₂ films were also investigated.

2. Experimental procedure

Chelates of copper, yttrium and calcium with ligands of 2,2,6,6-tetramethyl-3,5-heptanedione (thd) were used as precursors. Y-doped CaCuO₂ films were deposited on single crystals of SrTiO₃ (100). The deposition rate was controlled by the sublimation temper-

atures of the chelates (the sublimation temperatures of $Cu(thd)_2$, $Y(thd)_3$ and $Ca(thd)_2$ were varied in the range 95-105 °C, 160-170 °C and 180-190 °C, respectively). The film preparation was carried out at the total pressure of 100-200 Pa. The thicknesses of the films were measured by means of a surface roughness meter (Kosaka SE-30D). A film deposited for 1 h was about 50 nm thick. The composition of the metal elements in the films was determined by both the atomic absorption spectrometry and the inductively coupled plasma-emission spectrometry of a solution, which was prepared by dissolving a film in a HNO₃ solution. The temperature dependence of the resistivity of a film was measured by a direct current (d.c.) four-probe method. The Seebeck effect was measured in the range 100-300 K.

3. Results

Y-doped CaCuO₂ films were formed at a suitable substrate temperature (730-790 °C) and at an optimum oxygen flow rate, as shown in Fig. 1. Ydoped CaCuO₂ films were formed in a narrow oxygen-flow-rate range at low temperatures and in a wide range of oxygen flow rates at fairly high temperatures. In addition, the optimum flow rate of oxygen increased with increasing substrate temperature (the accurate partial pressure of oxygen near the SrTiO₃ substrate was not measured.). The preferable flow rate of oxygen may be governed by the following equilibrium: 2CuO \rightarrow Cu₂O + 1/2O₂.

In order to prepare Y-doped CaCuO₂ films (x > 0.1), the molar ratio of vaporized Y(thd)₃ to Cu(thd)₂ was enhanced up to 0.2. However, the formation of Ca_{1-x}Y_xCuO₂ was not confirmed at a vaporized molar ratio, Y(thd)₃/Cu(thd)₂, of 0.05 or more.



Figure 1 The relationship between the oxygen flow rate and the substrate temperature in the deposition processes of films: (\bigcirc) the formation of Y-doped CaCuO₂ confirmed by XRD, and (×) no formation of CaCuO₂.

TABLE I The composition of $Ca_{1-x}Y_xCuO_2$ films deposited on SrTiO₃ (100) substrates

Sample number (c-axis)	Substrate temper- ature (°C)	Vaporized precursors (mol) (Deposited metal elements (mol))		
		Cu	Y	Ca
1(0.3183 nm)	734	6.1×10^{-6}	3.0×10^{-8}	7.1 × 10 ⁻⁶
		(3.4×10^{-7})	(4.1×10^{-9})	(4.1×10^{-7})
2(0.3183 nm)	750	6.5×10^{-6}	1.5×10^{-8}	6.9 × 10 ⁻⁶
		(1.9×10^{-7})	(2.3×10^{-9})	(2.4×10^{-7})
3(0.3180 nm)	750	6.8×10^{-6}	3.0×10^{-8}	7.1×10^{-6}
		(2.1×10^{-7})	(4.1×10^{-9})	(2.2×10^{-7})
4(0.3183 nm)	770	6.5×10^{-6}	1.5×10^{-8}	6.6×10^{-6}
		(1.4×10^{-7})	(2.9×10^{-9})	(1.8×10^{-7})
5(0.3182 nm)	780	1.1×10^{-5}	3.0×10^{-8}	1.0×10^{-5}
		(2.6×10^{-7})	(3.3 × 10 ⁻⁹)	(3.4×10^{-7})

For several films exhibiting intense reflections of $CaCuO_2$, the number of moles of vaporized chelates and of the metal elements contained in these films are listed in Table I. As seen in Table I, the content of Y atoms is less than a few per cent, which is approximately close to the vaporized molar ratio, Y(thd)₃/Cu(thd)₂.

Fig. 2 shows the X-ray diffraction (XRD) patterns of the Y-doped CaCuO₂ film labelled 4. The XRD pattern is almost the same as in CaCuO₂, except for a slight difference in 2θ [3]. Thus, reflections at 2θ of 28.0 and 57.9 can be indexed to (001) and (002) of Ydoped CaCuO₂. Thus, the *c*-axis of the film is preferentially oriented perpendicular to the substrate surface. The c-axis of the Y-doped CaCuO₂ film labelled 4 is estimated to be 0.3183 nm. The c-axis values of the other films range from 0.3180 to 0.3183 nm. These values for the c-axis are slightly shorter than those of non-doped CaCuO₂ (c = 0.3188 nm) [3]. Thus, the Ydoping of CaCuO₂ results in shortening of the *c*-axis. Such shortening of the *c*-axis by electron doping has been observed in n-type superconductors, for example $Nd_{2-x}Ce_{x}CuO_{4}$. Information on the *a*-axis of Y-



Figure 2 XRD of the Y-doped CaCuO₂ film labelled 4 in Table I (CuK_{α}).



Figure 3 Plots of absolute values of the Seebeck coefficient against temperature for the Y-doped $CaCuO_2$ film labelled 4 in Table I. The Seebeck coefficient is negative.

doped $CaCuO_2$ could not be obtained because the (110) and (220) reflections of $CaCuO_2$ were not observed in XRD of films deposited on $SrTiO_3$ (110) substrates.

All Y-doped CaCuO₂ films prepared in the present work exhibit negative Seebeck coefficients at temperatures lower than 270 K. In Fig. 3, the Seebeck coefficient for the film labelled 4 is plotted as a function of temperature. The Seebeck coefficient is negative in the temperature range 110-270 K, so the carriers of these Y-doped CaCuO₂ films are electrons, which may be generated by the substitution of Y atoms for Ca sites in CaCuO₂. It should be noted that the Seebeck coefficient is nearly constant in the range of 270-100 K.

Fig. 4 shows plots of the resistivity, ρ , against the temperature, T, of the Y-doped CaCuO₂ films labelled 3 and 4. For these films, the temperature dependence of the resistivity is weak at temperatures higher than 50 K, whereas the resistivity increases remarkably at temperatures higher than 50 K. Unfortunately, no drastic decrease in the resistivity is observed for films 3 and 4 at temperatures higher than 12 K; no superconductivity is found for the Y-doped CaCuO₂ films at temperatures higher than 12 K. The temperature dependence of the resistivity shown in Fig. 4 has been observed for lightly doped La_{2-x}Sr_xCuO₂ (x < 0.06)



Figure 4 Plots of ρ against T for the Y-doped CaCuO₂ films labelled 3 and 4 in Table I.

and superconductivity emerged with further hole-doping [4]. Thus, superconductivity might appear for a Ydoped CaCuO₂ film with a higher carrier concentration if the problem of the low solubility of Y atoms in CaCuO₂ is overcome.

4. Discussion

As seen in Table I, the Y-content in $CaCuO_2$ is a few per cent. This result implies that the solubility limit of Y-atoms in $CaCuO_2$ is of the order of a few per cent. The ionic radius of Y^{3+} ions is close to that of Ca^{2+} ions, so that the substitution of Y^{3+} ions for Ca^{2+} ions is expected to occur to some extent, from a crystallographic viewpoint. In fact, superconductivity was found at 65 K for $Pb_2Sr_2Y_{1-x}Ca_xCu_3O_8$ (x = 0.5) [5]. Thus, the low solubility of Y atoms in $CaCuO_2$ may be due to the instability of $CaCuO_2$, because $CaCuO_2$ is not synthesized at 1 atm.

The Seebeck coefficient, S, for non-degenerate semiconductors is written as $S = (k/q)(5/2 + r + E_f/kT)$, where $E_{\rm F}$ is the Fermi energy with respect to the bottom of the conduction band and r is a constant dependent on the scattering mechanism. The temperature independence of the Seebeck coefficient shown in Fig. 3 means the zero-Fermi-energy: carriers of Ydoped CaCuO₂ are degenerate. In general, the zero Fermi energy is realized when the carrier concentration is larger than the effective state density of the conduction band. Under this condition, the Fermi level is located in the conduction band and thus the carrier concentration is independent of temperature. As seen in Fig. 4, however, the resistivity is remarkably high near 50 K. Such a temperature dependence of the resistivity is not typical of normal degenerate semiconductors. The localization of carriers occurs for conductors containing a large amount of the random potential (disordered conductors), even if an appreciable state density is present at the Fermi level. The localization of carriers is called Anderson localization



Figure 5 Plots of σ against ln T for the Y-doped CaCuO₂ films labelled 3 and 4 in Table I.

[6]. According to a recent theory of Anderson localization, the localization of carriers always occurs for the case of two-dimensional conductors, and occurs in the case of three-dimensional conductors with a carrier density less than a critical value. In the Anderson localization of a weakly localized regime, the conductivity consists of a term proportional to ln T for a twodimensional conductor [7]. For instance, a linear relationship was observed in plots of the conductivity, σ , against ln T for Nd_{2-x}Ce_xCuO₄ films with low carrier densities [8]. In Fig. 5, the conductivity, σ , of films 3 and 4 is plotted as a function of $\ln T$. No linear relationship is seen in Fig. 5. Accordingly, the localization of carriers of the Y-doped CaCuO₂ films labelled 3 and 4 is not in accordance with a weakly localized regime of two-dimensional conductors. In the weakly localized regime of three-dimensional conductors, the conductivity contains a term proportional to temperature [9]. In Fig. 6, the conductivity, σ , is plotted as a function of T. In Fig. 6, it seems that the conductivity is proportional to T at temperatures lower than 180 K. However, zero conductivity at T = 0 is anticipated from an extrapolation of the σ versus T plot. Thus, the temperature dependence of the conductivity shown in Fig. 5 is not explained by the weak localization in three-dimensional conductors, in which a nonzero minimum conductivity should be obtained at T = 0.

Although an explicit formula for strong localization at finite temperatures is not presented, the temperature dependence of the conductivity can be qualitatively expressed by variable range hopping in which the conductivity, σ , is given by $\sigma(T) = \sigma_0$ $\exp[-(T_0/T)^{1/(d+1)}]$, where *d* is the dimensionality [10]. In Figs 7 and 8, ln σ is plotted as a function of $T^{-1/3}$ and $T^{-1/4}$, respectively. A good linear relationship is seen in the plots of ln σ against $T^{-1/4}$. A linear relationship has been observed in plots of ln σ against $T^{-1/3}$ for Nd_{2-x}Ce_xCuO₄ [8]. According to the theory of variable range hopping [10],



Figure 6 Plots of σ against T for the Y-doped CaCuO₂ films labelled 3 and 4.



Figure 7 Plots of $\ln \sigma$ against $T^{-1/3}$ for the Y-doped CaCuO₂ films labelled 3 and 4.

 $Nd_{2-x}Ce_xCuO_4$ is a two-dimensional conductor, whereas Y-doped CaCuO₂ is a three-dimensional conductor. This difference may be attributable to the lack of blocking layers between the CuO₂ sheets in the CaCuO₂ structure. Another explanation is attributed to the many defects in the *ab*-plane, because a linear relationship was found in plots of ln ρ against $T^{-1/4}$ for La_{2-x}Sr_xCuO₄ (x < 0.06) which has blocking layers between the CuO₂ sheets [4]; two-dimensional conduction is interrupted by defects in the *ab*-plane, and then the conduction along the *c*-axis is not neglected.

As mentioned above, strong localization occurs for disordered two-dimensional metals and for disordered



Figure 8 Plots of $\ln \sigma$ against $T^{-1/4}$ for the Y-doped CaCuO₂ films labelled 3 and 4.

three-dimensional metals with low carrier densities as the temperature is lowered. Besides these disordered metals, strong localization is expected for charge transport through an impurity band in heavily doped semiconductors [10]. As seen in Table I, if 2% of the Ca^{2+} ions in CaCuO₂ are substituted by Y³⁺ ions, the state density of the impurity levels due to Y atoms is estimated to be 4.2×10^{20} cm⁻³. This value of the state density is insufficient to form an impurity band over the CaCuO₂ crystal. Under this condition, variable range hopping is a predominant process at lower temperatures. The Y-doped CaCuO₂ films exhibit sharp reflection peaks in the XRD patterns, so that it seems that the Y-doped CaCuO₂ films are not corresponding to disordered metals. The origin of the strong localization of the Y-doped CaCuO₂ films may be related to the impurity band. Recently, the temperature dependence of $\ln \rho$ against $T^{-1/(d+1)}$ has been explained by the localization of bosons [11]. To explain the present work by bosons localization, it is necessary to measure the effects of the magnetic field on the resistivity.

References

- M. G. SMITH, A. MANTHIRAM, J. ZHOU, J. B. GOODE-NOUGH and J. T. MARKERT, Nature 351 (1991) 549.
- M. TAKANO, Y. TAKEDA, H. OKADA, M. MIYAMOTO and T. KUSAKA, *Physica C* 159 (1989) 375.
- 3. K. KOBAYASHI, Y. ISHIHARA, S. MATSUSHIMA and G. OKADA, Jpn. J. Appl. Phys. 30 (1991) L1931.
- 4. H. TAKAGI, T. ITO, S. ISHIBASHI, M. UOTA, S. UCHIDA and Y. TOKURA, *Phys. Rev. B* 40 (1989) 2254.
- R. J. CAVA, B. BATLOGG, J. J. KRAJEWSKI, L. W. RUPP, L. F. SCHNEEMEYER, T. SIEGRIST, R. B. VAN DOVER, P. MARSH, W. F. PECK Jr, P. K. GALLAGHER, S. H. GLARUM. J. H. MARSHALL, R. C. FARROW, J. V. WASZCZAK, R. HULL and P. TREVOR, *Nature* 336 (1988) 211.
- 6. E. ABRAHAMS, P. W. ANDERSON, D. C. LICCIAR-

DELLO and T. V. RAMAKRISHNAN, Phys. Rev. Lett. 42 (1979) 673.

- 7. L. P. GORKOV, A. Z. LAIRKN and D. E. KHMELNIT-ZKII, JETP Lett. 30 (1979) 248.
- 8. S. TANDA, M. HONMA and T. NAKAYAMA, *Phys. Rev. B* 43 (1991) 8725.
- 9. G. A. THOMAS and A. KAWABATA, *ibid.* 26 (1982) 2113.

N. F. MOTT, Rev. Mod. Phys. 50 (1978) 203.
M. P. A. FISHER, Phys. Rev. Lett. 65 (1990) 923.

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